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Chemical Reaction Engineering-I - Course

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Engineering 2 $\text{KClO}_3(\text{aq}) + 2 \text{KCl}(\text{aq}) \rightarrow$
 $3\text{O}_2(\text{g})$ $\text{CaCO}_3(\text{aq}) \rightarrow \text{CaO}(\text{aq}) + \text{CO}_2(\text{g})$

Combustion- A reaction where a
reactant burns in the presence of
oxygen to form carbon dioxide and

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water vapor. Note that here water is usually produced as vapor due to the high amount of heat energy produced in these types of reactions.

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