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Elements of Chemical Reaction Engineering  $2 \text{KClO}_3(\text{aq}) + 2 \text{KCl}(\text{aq}) \rightarrow 3 \text{O}_2(\text{g})$   $\text{CaCO}_3(\text{aq}) \rightarrow \text{CaO}(\text{aq}) + \text{CO}_2(\text{g})$  Combustion- A reaction where a reactant burns in the presence of oxygen to form carbon dioxide and water vapor. Note that here water is usually produced as vapor due to the high amount of heat energy produced in these types of reactions.

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